



## **Cambridge IGCSE**<sup>™</sup>

CANDIDATE NAME					
CENTRE NUMBER			CANDIDATE NUMBER		

**BIOLOGY** 0610/62

Paper 6 Alternative to Practical

May/June 2024

1 hour

You must answer on the question paper.

No additional materials are needed.

## **INSTRUCTIONS**

- Answer all questions.
- Use a black or dark blue pen. You may use an HB pencil for any diagrams or graphs.
- Write your name, centre number and candidate number in the boxes at the top of the page.
- Write your answer to each question in the space provided.
- Do **not** use an erasable pen or correction fluid.
- Do not write on any bar codes.
- You may use a calculator.
- You should show all your working and use appropriate units.

## **INFORMATION**

- The total mark for this paper is 40.
- The number of marks for each question or part question is shown in brackets [].

This document has 12 pages. Any blank pages are indicated.



A student investigated the effect of acid concentration on the rate of diffusion in agar jelly. The student was given three test-tubes containing bromothymol blue indicator in agar jelly.

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Bromothymol blue is a blue indicator that turns yellow in the presence of an acid.

The student used this method:

- Step 1 label three beakers A1, A2 and A3
- Step 2 put 20 cm<sup>3</sup> of 1 mol per dm<sup>3</sup> hydrochloric acid solution into beaker A1
- Step 3 put 2cm<sup>3</sup> of 1 mol per dm<sup>3</sup> hydrochloric acid solution into beaker **A2**
- Step 4 add 18 cm<sup>3</sup> of distilled water to the hydrochloric acid solution in beaker **A2**
- Step 5 use a clean syringe to transfer 2cm<sup>3</sup> of the solution from beaker **A2** to beaker **A3**
- Step 6 add 18 cm<sup>3</sup> of distilled water to beaker A3

Table 1.1 shows the final concentrations of the hydrochloric acid solution in each beaker.

Table 1.1

beaker	final concentration of the hydrochloric acid solution/mol per dm <sup>3</sup>	final volume of the solution/cm <sup>3</sup>
<b>A</b> 1	1.00	20
A2	0.10	
А3	0.01	20

(a) (i) Complete Table 1.1 by calculating the final volume of the solution in beaker A2.

Write your answer in Table 1.1.

[1]

- Step 7 label the three test-tubes that contain the bromothymol blue indicator in agar jelly **A1**, **A2** and **A3**
- Step 8 use a syringe to transfer 10 cm<sup>3</sup> of the hydrochloric acid solution from beaker **A1** to test-tube **A1**
- Step 9 use a syringe to transfer 10 cm<sup>3</sup> of the hydrochloric acid solution from beaker **A2** to test-tube **A2**
- Step 10 use a syringe to transfer 10 cm<sup>3</sup> of the hydrochloric acid solution from beaker A3 to test-tube A3
- Step 11 place the three test-tubes into a water-bath at 35 °C
- Step 12 start a stop-clock and leave the test-tubes in the water-bath for 20 minutes
- Step 13 after 20 minutes, remove the test-tubes from the water-bath
- Step 14 measure the temperature of the water in the water-bath.



Fig. 1.1 shows the three test-tubes after they were removed from the water-bath in step 13.

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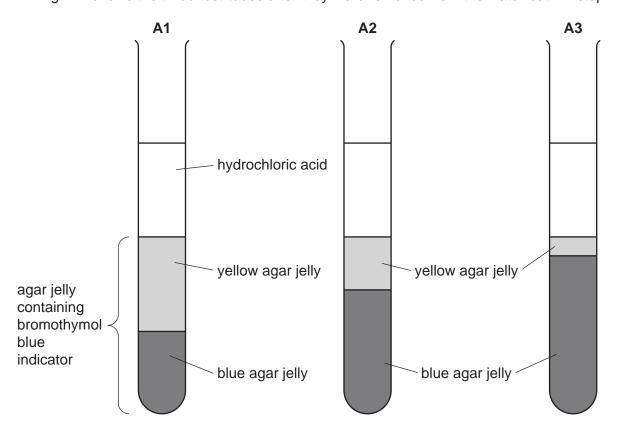


Fig. 1.1

(ii) Measure the height of the yellow agar jelly in each test-tube shown in Fig. 1.1.

Prepare a table and record the results of the investigation. Include in your table:

- your measurements from Fig. 1.1
- the final concentration of hydrochloric acid.

[2]

State a conclusion for the investigation.

	[1]
(iv)	Identify the independent variable and the dependent variable in this investigation.
	independent variable
	'
	dependent variable

Fig. 1.2 shows the reading on the thermometer in step 14.

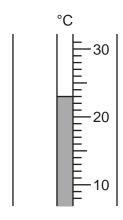


Fig. 1.2

Record the temperature of the water in the water-bath at the end of the investigation, shown in Fig. 1.2.

[1]

The student stated that the change in the temperature of the water-bath was not a significant source of error in the investigation.

Explain why the student was correct to state this.



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(vii)	Suggest a control experiment for this investigation.
	[1]
(viii)	The student repeated the investigation but left the test-tubes in the water-bath for 30 minutes instead of 20 minutes.
	Suggest how the results for this second investigation would differ from the results for the first investigation.
	[1]

**(b)** In an experiment, a student cut different-sized cubes of agar containing bromothymol blue indicator. The student placed the cubes in hydrochloric acid and measured the time taken for each cube to become yellow.

The results are shown in Table 1.2.

Table 1.2

length of the cube sides/mm	volume of the cube/mm <sup>3</sup>	time taken for the cubes to become yellow/s				
2	8	35				
4		60				
6	216	170				
8	512	275				
10	1000	445				

(i) Complete Table 1.2 by calculating the volume of the 4mm cube.

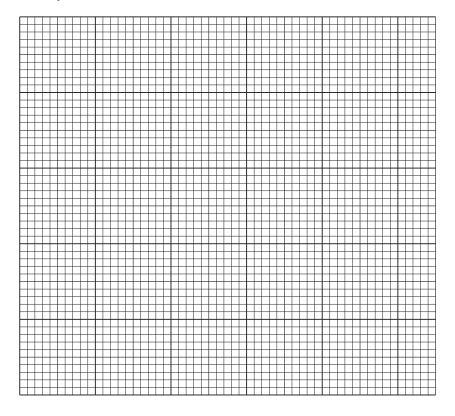
Write your answer in Table 1.2.

[1]



\* 0019655434707 \* (ii) Plot a l cubes \*

(ii) Plot a line graph on the grid of the length of the cube sides against the time taken for the cubes to become yellow.



[Total: 21]

[4]

(a) When a seed germinates it grows into a seedling.

Fig. 2.1 is a photograph of a seedling.



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Fig. 2.1

i) Make a large drawing of the seedling shown in Fig. 2.1.





(ii) Line AB represents the width of the seedling.

Measure the length of line AB on Fig. 2.1.

length of line AB .....mm

The actual width of the seedling is 9mm.

Calculate the magnification of the seedling in Fig. 2.1 using the formula and your measurement.

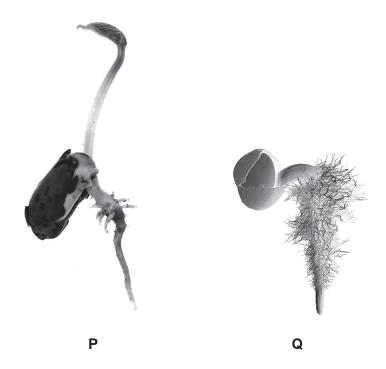
 $magnification = \frac{length of line$ **AB** $in Fig. 2.1}{actual width of the seedling}$ 

Give your answer to two significant figures.

Space for working.



Fig. 2.2 shows photographs of two different seedlings, **P** and **Q**.



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Fig. 2.2

	seedling <b>Q</b> in Fig. 2.2.
	similarity 1
	difference 1
	difference 2
	[3]
b)	As seedlings grow, they change starch into reducing sugars.
	Describe the methods you would use to test a sample of seedlings for starch and reducing sugars.
	starch test
	reducing sugar test

\* 0019655434711 \*

(c) When light shines from one side, seedlings grow towards the light.

Fig. 2.3 shows a diagram of a seedling growing towards a light source. The angle of growth is shown on the diagram.

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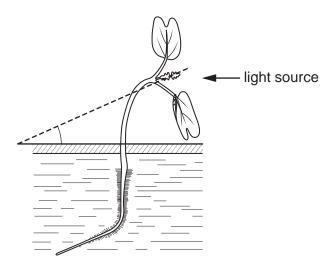


Fig. 2.3

seedling	g gr	ows.				intensit		-		
			 	 	 	 	 		 	[6]

[Total: 19]

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